## **Organic Chemistry**



*Recommended reading for this topic:* 

Lesson II.6-9 in *Organic Chemistry 1 Primer 2018*, by Rhett C. Smith, Andrew G. Tennyson and Tania Houjeiry

Additional Videos and how to match videos to your course text book: ProtonGuru.com



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An E1 reaction occurs when 2-bromo-2-methylpropane is heated in water. Provide a reasonable arrow-pushing mechanism for this reaction and label the final major neutral organic product that one would isolate after reaction.



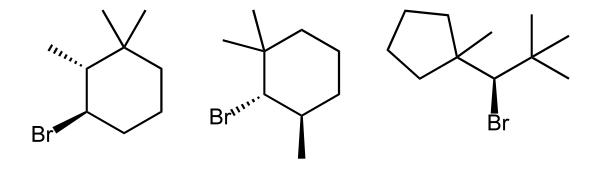
An E1 reaction occurs when 3-bromo-2,2-dimethylbutane is heated in water. Provide a reasonable arrow-pushing mechanism for this reaction and label the final neutral major organic product that one would isolate after reaction.



An E2 reaction occurs when 2-bromo-3-methylbutane is heated with  $NaOC_2H_5$ . Provide a reasonable arrow-pushing mechanism for this reaction and label the final neutral major organic product that one would isolate after reaction.



Which of these potential E2 substrates lack the antiperiplanar leaving group/proton pair that is required to be able to do an E2 reaction?





There are two possible E2 products for each substrate below. For each pair of reactions, determine which will produce more of the non-Zaitsev (Hofmann\_ product and explain your reasoning.

