

## ORGANIC CHEMISTRY 2 LECTURE GUIDE 2019

BY RHETT C. SMITH, PH.D.

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Executive Editor: Rhett C. Smith, Ph.D. You can reach him through our office at:

[IQ@protonguru.com](mailto:IQ@protonguru.com)

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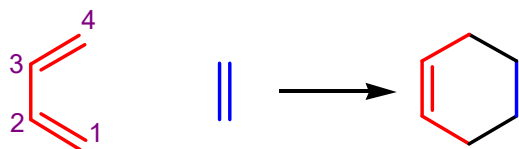
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**Lesson IV.3. Diels-Alder Reaction***A concerted cycloaddition reaction*

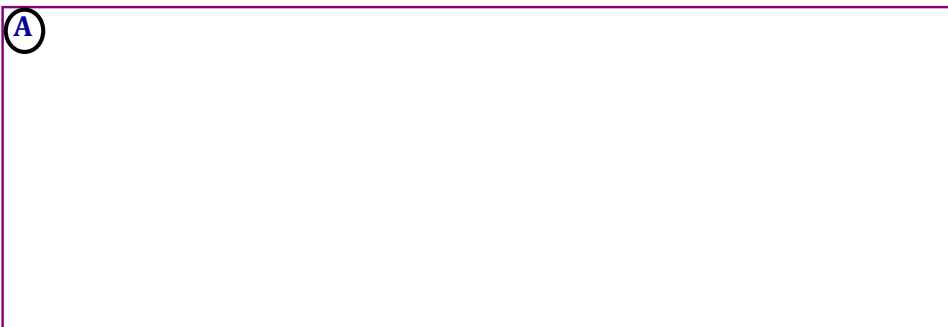
The Diels-Alder reaction is a [4+2] cycloaddition reaction; it makes a six-membered ring from a diene that supplies 4 atoms and a dienophile that supplies 2 atoms:



This reaction can be classified as a 1,4-addition, because the dienophile adds to the 1 and 4 positions of the diene.

*diene*      *dienophile*

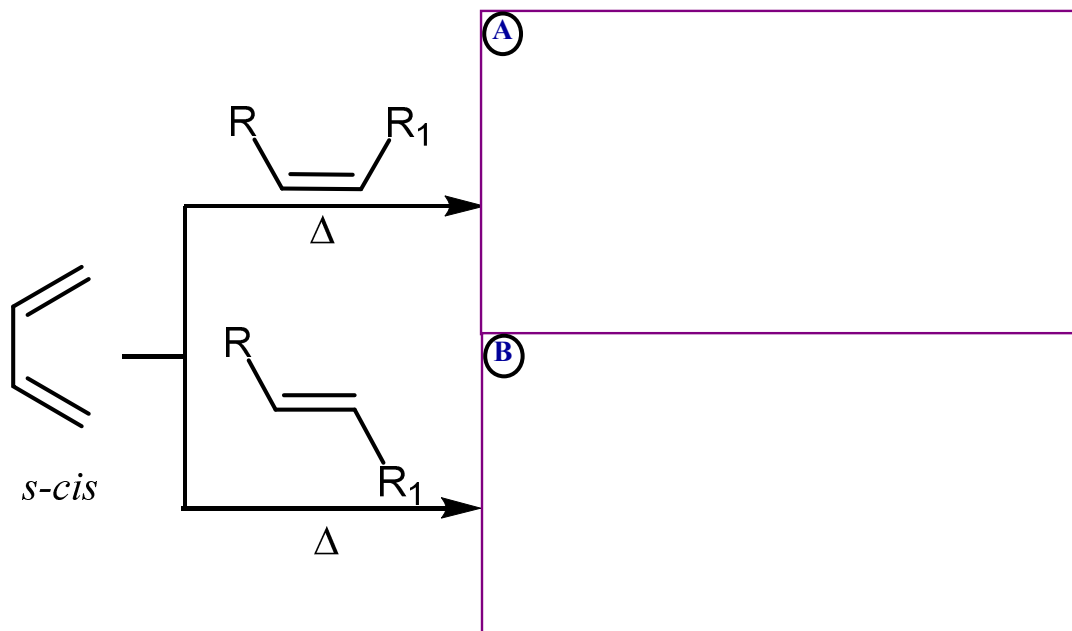
The mechanism of this reaction is very simple and it is a concerted, pericyclic reaction:



Notes

**Lesson IV.3. Diels-Alder Reaction***Regiospecificity of Diels-Alder Reactions*

One result of the concerted mechanism is that there is no rearrangement of starting materials in the course of the reaction, so the reaction is stereospecific:

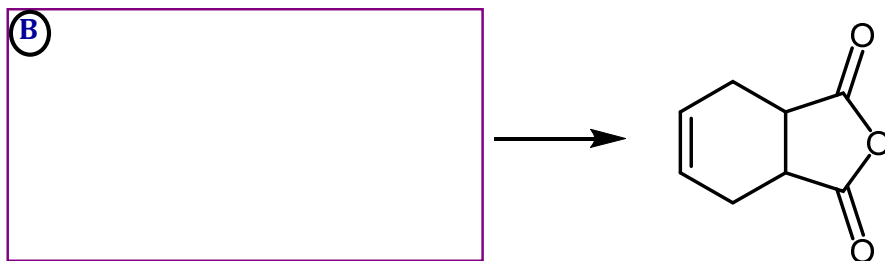
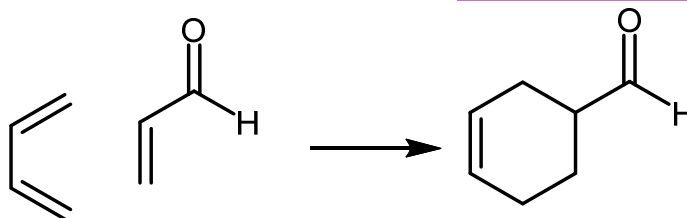
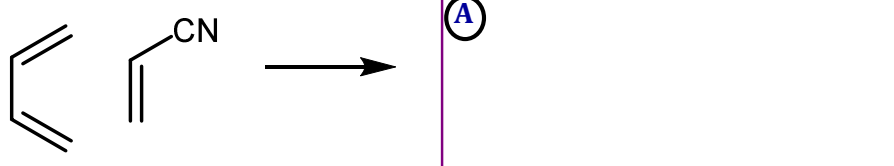


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Notes

**Lesson IV.3. Diels-Alder Reaction***Electron-Withdrawing Group Effect on Diels-Alder Reaction Rate*

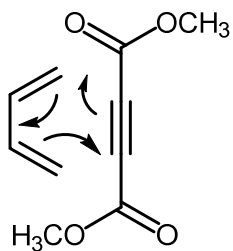
Diels -Alder reaction is facilitated by electron-poor dienophiles (dienophile substituted with an electron-withdrawing group):



Notes

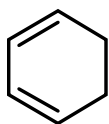
**Lesson IV.3. Diels-Alder Reaction***Alkynes and Cyclic Reagents for Diels-Alder Reactions*

The dienophile can also be an alkyne:

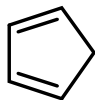


(A)

Cyclic compounds can be used as starting materials to create bridged bicyclic systems:



(B)



(C)

Notes